organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 298 KMean σ (C–C) = 0.005 Å R factor = 0.049 wR factor = 0.197 Data-to-parameter ratio = 15.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Cyclopentanone 4-nitrophenylhydrazone

The title compound, $C_{11}H_{13}N_3O_2$, includes a planar phenylhydrazone moiety and a five-membered carbon ring in an envelope conformation. A short C–C bond distance of 1.435 (6) Å is observed in the five-membered ring. This may be due to the larger displacement parameters of relevant atoms. Neighboring molecules are linked to each other *via* hydrogen bonding, forming a zigzag supramolecular chain along the *b* axis.

Comment

The molecular structure of (I) is shown in Fig. 1. The phenylhydrazone moiety is approximately planar. The fivemembered ring of the compound assumes an envelope conformation, with atom C10 at the flap position. The C9– C10 bond distance of 1.435 (6) Å is much shorter than the expected value for a normal Csp^3-Csp^3 bond, and is also



shorter than the average distance of 1.498 (5) Å for other C– C bonds in the same ring. The shortening of the C9–C10 bond may be due to the larger displacement parameters of both atoms C9 and C10. A similar short C–C bond of 1.448 Å was found in a cyclopentanone derivative, in which the relevant C atoms also had larger displacement parameters (Macias *et al.*, 1989). Neighboring molecules are linked to each other *via* both classical N–H···O hydrogen bonding and weak C– H···O hydrogen bonding, forming a zigzag supramolecular chain along the crystallographic *b* axis, as shown in Fig. 2.



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The structure of (I), shown with 30% probability displacement ellipsoids.

Received 30 September 2002 Accepted 7 October 2002

Online 18 October 2002

Experimental

Cyclopentanone (0.17 g, 0.2 mmol) and acetic acid (2 ml) were added in turn to an ethanol solution (20 ml) of 4-nitrophenylhydrazine (0.30 g, 0.2 mmol). After refluxing the solution for 40 min, it was filtered and the filtrate allowed to stand at room temperature. Yellow powder crystals appeared after 1 d. Well-shaped single crystals were obtained by recrystallization from chloroform.

> Mo $K\alpha$ radiation Cell parameters from 25 reflections $\theta = 4.4-12.1^{\circ}$ $\mu = 0.09 \text{ mm}^{-1}$ T = 298 (2) KPrism, orange $0.58 \times 0.46 \times 0.28 \text{ mm}$

 $h = -10 \rightarrow 26$

 $k = -7 \rightarrow 18$

3 standard reflections

every 150 reflections

intensity decay: 0.2%

 $l = -8 \rightarrow 3$

Crystal data

CHNO
$C_{11}\Pi_{13}\Pi_{3}O_{2}$
$M_r = 219.24$
Orthorhombic, Pbca
a = 21.372 (3) Å
b = 14.705 (4)Å
c = 7.211(2) Å
$V = 2266.2 (9) \text{ Å}^3$
Z = 8
$D_{\rm x} = 1.285 {\rm Mg}{\rm m}^{-3}$

Data collection

Rigaku AFC-7*S* diffractometer $\omega/2\theta$ scans 2360 measured reflections 2224 independent reflections 789 reflections with $I > 2\sigma(I)$ $R_{int} = 0.057$ $\theta_{max} = 26.0^{\circ}$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0862P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.049$	+ 0.1882P]
$wR(F^2) = 0.197$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{\rm max} < 0.001$
2224 reflections	$\Delta \rho_{\rm max} = 0.25 \ {\rm e} \ {\rm \AA}^{-3}$
145 parameters	$\Delta \rho_{\rm min} = -0.14 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

N1-C4	1.436 (4)	C7-C11	1.505 (5)
N2-C1	1.364 (4)	C8-C9	1.513 (5)
N2-N3	1.390 (3)	C9-C10	1.435 (6)
N3-C7	1.280 (4)	C10-C11	1.492 (6)
C7-C8	1.483 (5)		
C1-N2-N3	119.7 (2)	C7-N3-N2	115.6 (3)

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$N2-H2A\cdotsO1^{i}$	0.86	2.20	3.004 (4)	155
$C2-H2\cdots O1^i$	0.93	2.53	3.287 (4)	139

Symmetry code: (i) $\frac{1}{2} - x$, $y - \frac{1}{2}$, z.



Figure 2

A packing diagram for (I), showing the intermolecular hydrogen bonding.

H atoms were placed in calculated positions, with C–H = 0.93–0.97 Å and N–H = 0.86 Å. All H atoms were included in the final cycles of refinement in the riding mode, with isotropic displacement parameters of $1.2U_{eq}$ of the carrier non-H atoms.

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1992); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1985); program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994).

This project was supported by the National Natural Science Foundation of China (No. 29973036).

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